

N,N'-Bis(2-aminobenzyl)ethane-1,2-diaminium bis(4-methylbenzene-sulfonate)

Luis Ángel Garza Rodríguez,^a‡ Sylvain Bernès,^b Perla Elizondo Martínez,^a Blanca Nájera Martínez^a and Sara L. Rodríguez de Luna^{a*}

^aLaboratorio de Química Industrial, CELAES, Facultad de Ciencias Químicas, UANL, Pedro de Alba S/N, 66451 San Nicolás de los Garza, NL, Mexico, and ^bDEP Facultad de Ciencias Químicas, UANL, Guerrero y Progreso S/N, Col. Treviño, 64570 Monterrey, NL, Mexico

Correspondence e-mail: sylvain_bernes@hotmail.com

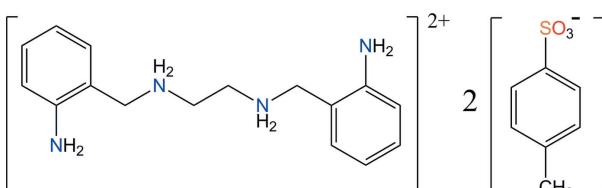
Received 24 October 2011; accepted 31 October 2011

Key indicators: single-crystal X-ray study; $T = 298\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$; R factor = 0.056; wR factor = 0.162; data-to-parameter ratio = 13.0.

The title salt, $\text{C}_{16}\text{H}_{24}\text{N}_4^{2+} \cdot 2\text{C}_7\text{H}_7\text{O}_3\text{S}^-$, crystallizes with the dication situated on an inversion center and the anion in a general position. The cation contains two ammonium and two free amine groups, and the observed conformation for the chain linking the benzene rings is different from that found in the free tetraamine and in the fully protonated tetraamine. All amine and ammonium H atoms of the cation form hydrogen bonds with eight symmetry-related anions, using the sulfonate O atoms as acceptors. This arrangement for the ions precludes any $\pi-\pi$ contacts between benzene rings in the crystal.

Related literature

For reviews on applications of macrocyclic systems, see: Vigato & Tamburini (2004); Radecka-Paryzek *et al.* (2005). For their acid-catalysed synthesis using *p*-toluenesulfonic acid, see: Ionkin *et al.* (2008). For the structures of the free molecule and the fully protonated cation corresponding to the title cation, see: Rodríguez de Barbarín *et al.* (2007) and Garza Rodríguez *et al.* (2009, 2011), respectively.



‡ Current affiliation: Universidad Regiomontana, Monterrey, NL, Mexico.

Experimental

Crystal data

$\text{C}_{16}\text{H}_{24}\text{N}_4^{2+} \cdot 2\text{C}_7\text{H}_7\text{O}_3\text{S}^-$	$\gamma = 97.80 (3)^\circ$
$M_r = 614.76$	$V = 753.6 (5)\text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 1$
$a = 5.753 (2)\text{ \AA}$	Mo $K\alpha$ radiation
$b = 9.512 (3)\text{ \AA}$	$\mu = 0.23\text{ mm}^{-1}$
$c = 14.493 (5)\text{ \AA}$	$T = 298\text{ K}$
$\alpha = 101.40 (2)^\circ$	$0.60 \times 0.16 \times 0.16\text{ mm}$
$\beta = 100.06 (3)^\circ$	

Data collection

Siemens P4 diffractometer	2234 reflections with $I > 2\sigma(I)$
Absorption correction: ψ scan (<i>XSCANS</i> ; Siemens, 1996)	$R_{\text{int}} = 0.091$
$T_{\text{min}} = 0.512$, $T_{\text{max}} = 0.594$	2 standard reflections every 98 reflections
3505 measured reflections	intensity decay: 1%
2650 independent reflections	

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.056$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.162$	$\Delta\rho_{\text{max}} = 0.39\text{ e \AA}^{-3}$
$S = 1.29$	$\Delta\rho_{\text{min}} = -0.34\text{ e \AA}^{-3}$
2650 reflections	
204 parameters	
4 restraints	

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D\cdots H\cdots A$	$D\cdots H$	$H\cdots A$	$D\cdots A$	$D\cdots H\cdots A$
$\text{N}1\cdots \text{H}11\cdots \text{O}2^{\text{i}}$	0.90 (1)	2.12 (1)	3.012 (3)	177 (3)
$\text{N}1\cdots \text{H}12\cdots \text{O}3^{\text{ii}}$	0.91 (1)	2.27 (3)	3.028 (4)	141 (3)
$\text{N}8\cdots \text{H}81\cdots \text{O}3^{\text{iii}}$	0.91 (1)	1.91 (2)	2.763 (3)	157 (3)
$\text{N}8\cdots \text{H}82\cdots \text{O}1^{\text{iv}}$	0.91 (1)	1.86 (1)	2.739 (3)	160 (3)

Symmetry codes: (i) $x - 1, y + 1, z$; (ii) $x, y + 1, z$; (iii) $-x + 1, -y + 1, -z + 2$; (iv) $-x + 2, -y + 1, -z + 2$.

Data collection: *XSCANS* (Siemens, 1996); cell refinement: *XSCANS*; data reduction: *XSCANS*; program(s) used to solve structure: *SHELXTL-Plus* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL-Plus*; molecular graphics: *SHELXTL-Plus* and *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXTL-Plus*.

The authors thank the FCQ-UANL (project No. 03-6375-QMT-08-005) and PAICyT (project No. IT164-09) for financial support. LAGR acknowledges a grant from the CONACyT program "Dirección de Tesis entre la UANL y la University of Texas at Austin y/o Instituciones de Educación Superior de la ANUIES" (grant N-L.-2006-C09-32658).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FJ2471).

References

- Garza Rodríguez, L. Á., Bernès, S., Elizondo Martínez, P., Nájera Martínez, B. & Pérez Rodríguez, N. (2011). *Acta Cryst. E67*, o3237–o3238.
- Garza Rodríguez, L. A., Bernès, S., Nájera Martínez, B., Elizondo Martínez, P. & Pérez Rodríguez, N. (2009). *Acta Cryst. E65*, o2995.
- Ionkin, A. S., Marshall, W. J., Adelman, D. J., Bobik Fones, B., Fish, B. M., Schiffhauer, M. F., Soper, P. D., Waterland, R. L., Spence, R. E. & Xie, T. (2008). *J. Polymer Sci. Polymer Chem.* **46**, 585–611.

organic compounds

- Macrae, C. F., Edgington, P. R., McCabe, P., Pidcock, E., Shields, G. P., Taylor, R., Towler, M. & van de Streek, J. (2006). *J. Appl. Cryst.* **39**, 453–457.
- Radecka-Paryzek, W., Patroniak, V. & Lisowski, J. (2005). *Coord. Chem. Rev.* **249**, 2156–2175.
- Rodríguez de Barbarán, C., Bernès, S., Nájera, B., Elizondo, P. & Cerda, P. (2007). *Acta Cryst. E* **63**, o549–o550.
- Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
- Siemens (1996). *XSCANS*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Vigato, P. A. & Tamburini, S. (2004). *Coord. Chem. Rev.* **248**, 1717–2128.